## Tyrolobibenzyls – Novel Secondary Metabolites from Scorzonera humilis

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Tyrolobibenzyls A (2), B (3), and C (4) were isolated from *Scorzonera humilis* (Asteraceae) of Tyrolean origin. Compounds 2 and 3 possess a unique phenylethyl-benzofuran backbone and represent a new class of natural compounds. Structural assignments are based on 1D and 2D NMR as well as MS data. Furthermore, peracetylated derivatives of compounds 3 and 4 were synthesized. Pharmacological evaluation of these compounds by the [3H]thymidine assay showed no pronounced cytotoxicity.

**1. Introduction.** – *Scorzonera humilis* L. (Asteraceae) is a perennial rosette plant of 10-40 cm height with yellow solitary flowerheads. Its distribution area covers the submediterranean and temperate zone of Europe from Portugal to Russia [1]. In Central Europe, this species is restricted to wet habitats and is considered an endangered plant [2]. In traditional medicine, *S. humilis* has been used as a remedy for wound healing and gastro-intestinal disorders [3]. Knowledge about the secondary metabolite spectrum of the whole genus *Scorzonera* is limited. Prior investigations of *S. hispanica* (black salsify) led to the isolation and identification of 3,4-dimethoxycinnamic acid methyl ester,  $\beta$ -sitosterol, the lignan (1S,3aR,4S,6aR)-1,4-bis $(4-\beta$ -D-glucopyranosyloxy-3,5-dimethoxyphenyl)tetrahydrofuro[3,4-c]furan, the bisabolane derivative puliglutone, as well as the sesquiterpene lactone glucosides scorzoneroside, ixerisoside D, and 3-O-angeloyl-11 $\beta$ ,13-dihydrodesacylcynaropicrin-8-yl  $\beta$ -D-glucoside [4].

**2. Results and Discussion.** – The MeOH extract (60.0 g) of air-dried subaerial parts (190.8 g) of *S. humilis* was partitioned between  $H_2O/MeOH$  and AcOEt, and the AcOEt fraction was subjected to repeated column chromatography (CC) over silica gel and then over *Lobar RP18*, yielding compounds **1** (17.2 mg), **2** (133.8 mg), **3** (158.2 mg), and **4** (93.4 mg). Furthermore, compounds **3** and **4** were acetylated and subsequently purified by means of CC (silica gel) and semi-prep. HPLC to give the pure peracetyl derivatives **3a** and **4a**, respectively.

Compound **1** was identified by comparison of its ESI-MS and  $^{1}$ H- and  $^{13}$ C-NMR data with literature values as the lignan pinoresinol-1-yl  $\beta$ -D-glucopyranoside (=(1S,3aR,4R,6aR)-tetrahydro-1,4-bis(4-hydroxy-2-methoxyphenyl)-1H,3H-furo[3,4-c]furan-3a-yl  $\beta$ -D-glucopyranoside), formerly isolated from *Stauntonia hexaphylla* DECNE (Lardizabalaceae), *Forsythia suspensa* VAHL. (Oleacae) and *Olea europaea* L. as well as *Olea africana* MILL. (Oleacae) [5].

The ESI-MS of compound **2** displayed signals at m/z 443 ( $[M-H]^-$ ) and 281 ( $[M-glucose-H]^-$ ), which is consistent with the molecular formula  $C_{23}H_{24}O_9$ . The <sup>1</sup>H-NMR spectrum ( $Table\ I$ ) exhibited signals for a  $\beta$ -D-glucose moiety, signals assignable to a para-substituted aromatic ring, three vicinal aromatic protons and one isolated methine group (H-C(3)), indicating the presence of a second aromatic ring system, and signals for two methylene groups attributable to an ethanediyl moiety.

These data, together with the <sup>13</sup>C-NMR and DEPT spectra (*Table 2*) and the HSQC and HMBC experiments (*Table 3*) established the structure of **2** as 4-[2-(4-hydroxyphenyl)ethyl]benzofuran-2-carboxylic acid (2*S*,3*R*,4*S*,5*S*,6*R*)-tetrahydro-3,4,5-trihydroxy-6-(hydroxymethyl)-2*H*-pyran-2-yl ester (=  $\beta$ -D-glucopyranosyl 4-[2-(4-hydroxyphenyl)ethyl]benzofuran-2-carboxylate). This substance belongs to a new type of natural compounds for which we propose the name tyrolobibenzyls due to the origin of the plant material and one characteristic part of the molecule; thus, **2** is called tyrolobibenzyl A. So far neither natural nor synthesized (phenylethyl)benzofuran-2-carboxylic acid derivatives are known in the literature <sup>1</sup>).

The  $^{13}$ C-NMR and DEPT spectra ( $Table\ 2$ ) of **2** showed signals of thirteen CH groups, seven quarternary C-atoms, and three CH<sub>2</sub> groups. Two of the aromatic CH signals at  $\delta$ (C) 130.5 and 116.0 exhibited double intensities and could be assigned to the atoms C(2'), C(6'), and C(5') of a 4-substituted phenyl ring, respectively. Taking into account the downfield-shifted signal of C(4') ( $\delta$ (C) 156.4) and mass spectrometry data, the substituent at C(4') had to be a OH group.

HSQC and HMBC Experiments (*Table 3*) with **2** established the second aromatic system as a benzofuran-2-carboxylic acid derivative. Accordingly, the HMBC spectrum showed couplings between H-C(5) and C(3a), H-C(7) and C(3a), H-C(6) and C(7a), H-C(3) and C(3a), as well as H-C(3) and the exocyclic COO group.

The linkages between the 4-hydroxyphenyl ring, the benzofuran, and the sugar moiety of  $\mathbf{2}$  were deduced from the HMBC experiment, too. A cross-peak between the signal of H-C(1'') ( $\delta(H)=5.40$ ) and the signal of the exocyclic COO group ( $\delta(C')=168.7$ ) revealed that the anomeric OH group of  $\beta$ -D-glucose was esterified with the 2-carboxy moiety of the benzofuran ring. On the other hand, two- and three-bond couplings between the CH<sub>2</sub> groups of the ethanediyl moiety with C(6'), C(1'), and C(2'), as well as with C(4), C(5), and C(3a) confirmed that the 4-hydroxyphenyl ring was attached at C(4) of the benzofuran ring via CH<sub>2</sub>CH<sub>2</sub>.

The ESI-MS of compound 3 displayed signals at m/z 459 ([M-H]<sup>-</sup>) and 297 ([M-glucose -H]<sup>-</sup>), which is consistent with the molecular formula  $C_{23}H_{24}O_{10}$ . Except for differences due to a missing aromatic CH group at position 5, which in

<sup>1)</sup> Chemical Abstracts Services online search from April 28, 2000.

Table 1. <sup>1</sup>*H-NMR Spectra of* **2**, **3**, **3a**, **4**, and **4a**<sup>a</sup>)<sup>b</sup>).  $\delta$  in ppm rel. to SiMe<sub>4</sub>, J in Hz.

	2	3	3a <sup>c</sup> )		4	<b>4a</b> <sup>d</sup> )
H-C(3)	6.04 (s)	6.02 (s)	5.89 (s)	H-C(4)	7.21	7.07
					(d, J(4,5) = 9.0)	(d, J(4,5) = 8.5)
H-C(5)	6.96			H-C(5)	6.73	6.99
	(dd, J(5,6) = 8.0,				(d, J(5,4) = 9.0)	(d, J(4,5) = 8.5)
	J(5,7) = 1.0					
H-C(6)	7.41	7.14	7.25 (d, J = 9.0)	MeCO-C(1)	2.33 (s)	2.28(s)
	(dd, J(6,5) = 8.0,	(br. d, J = 9.0)				
	J(6,7) = 8.0					
H-C(7)	7.18	7.10	7.21 (d, J = 9.0)			
	(dd, J(7,6) = 8.0,	(br. d, J = 9.0)				
	J(7,5) = 1.0					
$2H-C(\alpha)$	3.75(m)	3.58 (m)	3.37	$2H-C(\alpha)$	2.91 (m)	$2.70 \ (m^{\rm f}))$
	244 ( )	2.22 ( )	(dt, J = 13.5, 7.0)		200()	
	3.11 (m)	3.33 (m)	2.90		2.80 (m)	
211 ((4)	200()	2.76 ( )	(dt, J = 13.5, 7.0)	211 ((4)	200()	2.01 ( )
$2H-C(\beta)$	2.89 (m)	2.76 (m)	2.90	$2H-C(\beta)$	2.80 (m)	2.81 (m)
	2.77 ()		(dt, J = 13.5, 7.0) 2.80		2.80 (m)	2.71 (f))
	2.77(m)		(dt, J = 13.5, 7.0)		2.80 (m)	$2.71 \ (m^{\rm f}))$
H-C(2')	6.97°), 6.98°)	7.15°), 7.14°)	(ui, J = 13.3, 7.0) $7.04^{\circ}), 7.03^{\circ})$	H-C(2')	7.06°), 7.05°)	7.18°), 7.17°)
H-C(2) H-C(3')	6.66°), 6.67°)	6.73°), 6.72°)	6.91°), 6.90°)	H-C(2) H-C(3')	6.75°), 6.74°)	6.98°), 6.97°)
H-C(5')	6.66°), 6.67°)	6.73°), 6.72°)	6.91°), 6.90°)	H-C(5')	6.75°), 6.74°)	6.98°), 6.97°)
H-C(6')	6.97°), 6.98°)	7.15°), 7.14°)	7.04°), 7.03°)	H-C(6')	7.06°), 7.05°)	7.18°), 7.17°)
H-C(1")	$5.40 (d, {}^{3}J = 8.0)$	5.33	5.44 (d, J = 8.0)	H-C(1")	4.90 (d, J = 8.0)	5.12 (d, J = (8.0))
11 ((1)	5.10 (a, 5 = 0.0)	(d, J = 8.0)	5.11 (a, 5 = 0.0)	11 ((1)	1.50 (a, 5 = 0.0)	5.12 (a, 5 – (6.6)
H-C(2")	$3.59 (m^{\rm f})$	3.62	5.23	H-C(2")	3.57	5.38
(- )	( )	(dd, J = 8.0, 8.0)	(dd, J = 9.5, 8.0)	(- )	(dd, J = 9.5, 8.0)	(dd, J = 9.5, 8.0)
H-C(3")	3.51	3.53	5.32	H-C(3")	3.52 (m)	5.31
` '	$(dd, {}^{3}J = 9.0, 9.0)$	(dd, J = 9.0, 9.0)	(dd, J = 9.5, 9.5)	. ,	. ,	(dd, J = 9.5, 9.5)
H-C(4")	3.44	3.40 (m)	5.16	H - C(4'')	3.46(m)	5.21
	$(dd, {}^{3}J = 9.0, 9.0)$		(dd, J = 9.5, 9.5)			(dd, J = 9.5, 9.5)
H-C(5")	3.57 (m <sup>f</sup> )	3.48	3.93	H-C(5")	3.46 (m)	3.87
		(br. d, J = 9.0)	(ddd, J = 9.5, 6.0, 2.0)			(ddd, J = 9.5, 5.5, 2.5)
2H-C(6")	3.94	3.96	4.30	2H-C(6'')	3.96	4.29
	$(dd, {}^{2}J = 12.5, {}^{3}J = 2.5)$	(dd, J = 12.5, 2.5)	(dd, J = 12.5, 6.0)		(dd, J = 12.0, 1.5)	(dd, J = 12.5, 5.5)
	3.75	3.78	4.19		3.77	4.18
	$(dd, {}^{2}J = 12.5, {}^{3}J = 5.5)$	(dd, J = 12.5, 5.5)	(dd, J = 12.5, 2.0)		(dd, J = 12.0, 5.0)	(dd, J = 12.5, 2.5)

a) All assignments are based on HSQC and HMBC experiments. b) Arbitrary numbering (see Formulae). c) MeCO s at 2.28, 2.20, 2.10, 2.08, 2.03, and 2.00. d) MeCO s at 2.29, 2.26, 2.06 (2×), 2.04, 1.96. c) Most intense signals of the AA'XX' spin system. f) Overlapping signals.

compound **3** is replaced by a quaternary C-atom substituted by an OH group.  $^1$ H-( $Table\ 1$ ) and  $^{13}$ C-NMR ( $Table\ 2$ ) data of **3** were nearly identical with those of **2**. Hydroxylation at C(5) was supported by comparison of the  $^{13}$ C-NMR spectrum of **3** with that of **2** (downfield-shifted signal of C(5) ( $\Delta \delta = +24.2$  ppm) and upfield-shifted signals for the vicinal C(4) and C(6) ( $\Delta \delta = 15.2$  and -12.0 ppm, resp.) in **3**). The assigned structure was confirmed by HSQC and HMBC experiments with **3** ( $Table\ 3$ ) and the data of peracetylated derivative **3a** ( $Table\ 1$  and 2). Thus, **3** is 5-hydroxy-4-[2-(4-hydroxyphenyl)ethyl]benzofuran-3-carboxylic acid (2S, 3R, 4S, 5S, 6R)-tetrahydro-3,4,5-trihydroxy-6-(hydroxymethyl)-2H-pyran-2-yl ester ( $=\beta$ -D-glucopyranosyl 5-hydroxy-4-[2-(4-hydroxyphenyl)ethyl]benzofuran-2-carboxylate) or tyrolobibenzyl B.

The ESI-MS of compound 4 displayed signals at m/z 433 ([M-H]<sup>-</sup>) and 271 ([M-glucose -H]<sup>-</sup>), which is consistent with the molecular formula  $C_{22}H_{26}O_9$ . Partial structures could be established by the <sup>1</sup>H-NMR spectrum (*Table 1*), which revealed

Table 2.  $^{13}C$ -NMR Spectra of **2**, **3**, **3a**, **4**, and **4a**. Signal assignments by HSQC and HMBC data,  $\delta$  in ppm rel. to SiMe<sub>4</sub>.

	2	3	3a <sup>a</sup> )		4	<b>4a</b> <sup>b</sup> )
C(2)	164.8	165.4	162.5	C(1)	131.0	136.4
CH(3)	94.3	94.2	95.1	C(2)	130.0	129.1
C(3a)	114.8	115.4	114.0	C(3)	150.5	152.8
C(4)	142.8	127.6	121.4	CH(4)	119.4	116.1
CH(5) or $O-C(5)$	129.5	153.7	151.7	CH(5)	114.7	121.5
CH(6)	133.0	121.0	129.5	C(6)	150.4	141.9
CH(7)	116.3	116.2	116.3	$Me_3CO$	208.4	202.7
C(7a)	156.2	149.5	154.8	MeCO	32.1	32.0
COO	168.7	169.2	165.9	$CH_2(\alpha)$	31.0	30.4
$CH_2(\alpha)$	39.2	30.8	27.8	$CH_2(\beta)$	36.6	35.7
$CH_2(\beta)$	38.5	36.7	35.7	C(1')	134.0	139.3
C(1')	133.8	135.0	138.8	CH(2')	129.7	129.7
CH(2')	130.5	130.4	130.0	CH(3')	116.0	121.7
CH(3")	116.0	116.0	121.7	C(4')	156.4	149.2
C(4')	156.4	156.2	149.5	CH(5')	116.0	121.7
CH(5')	116.0	116.0	121.7	CH(6')	129.7	129.7
CH(6')	130.5	130.4	130.0	CH(1")	103.8	99.2
CH(1")	101.2	101.1	96.8	CH(2")	75.3	71.2
CH(2")	74.4	74.3	72.6	CH(3")	78.5	73.0
CH(3")	78.3	78.7	71.2	CH(4")	71.5	68.4
CH(4")	70.9	70.9	67.7	CH(5")	78.2	72.3
CH(5")	78.8	78.3	72.8	$CH_2(6'')$	62.6	62.0
CH(6")	62.3	62.3	61.7			

<sup>&</sup>lt;sup>a</sup>) MeCO: 170.8, 21.2; 171.2, 21.0; 169.8, 20.7; 170.3, 20.7; 170.3, 20.6; 170.3, 20.6. <sup>b</sup>) MeCO: 170.6, 20.8; 170.3, 20.7; 169.9, 21.3; 169.8, 20.7; 169.5, 20.7; 169.3, 21.2.

Table 3. Important HMBC Cross-Peaks Observed for Compounds 2-4

2		4		4a	
Proton	C-Atom	Proton	C-Atom	Proton	C-Atom
H-C(3)	C(2), COO, C(3a)	H-C(4)	C(2), C(6)	H-C(4)	C(2), C(3), C(5), C(6)
H-C(5)	C(3a), C(4), C(7)	H-C(5)	C(1), C(3)	H-C(5)	C(1), C(3), C(4), C(6)
H-C(6)	C(4), C(5), C(7), C(7a)				. , , , ,
H-C(7)	C(3a), C(5), C(7a)	MeCO	MeCO	MeCO	MeCO, C(1)
$H_a$ - $C(\alpha)$	C(3a), C(4), C(5), C(1')	$H_a$ - $C(\alpha)$	$C(1), C(2), C(3), C(\beta), C(1')$	$H_a$ - $C(\alpha)$	$C(1), C(2),  C(3), C(\beta), C(1')$
$H_b-C(\alpha)$	C(3a), C(4), C(5), C(1')	$H_b-C(\alpha)$	$C(2), C(3), C(\beta), C(1')$	$H_b$ - $C(\alpha)$	$C(2), C(3),  C(\beta), C(1')$
$H_a-C(\beta)$	C(4), C(1'), C(2')/C(6')	$H_a$ - $C(\beta)$	C(2), $C(a)$ , $C(1')$	$H_a$ - $C(\beta)$	$C(2)$ , $C(\alpha')$ , $C(1')$
$H_b-C(\beta)$	C(4), C(1'), C(2')/C(6')	$H_b-C(\beta)$	$C(2)$ , $C(\alpha)$ , $C(1')$	$H_b-C(\beta)$	$C(2)$ , $C(\alpha')$ , $C(1')$
H-C(2')/H-C(6')	C(β), C(1'), C(2')/C(6'), C(3')/C(5'), C(4')	H-C(2')/H-C(6')	C(β), C(2')/ C(6'), C(4')	H-C(2')/ H-C(6')	C(β), C(2')/ C(6'), C(4')
H-C(3')/H-C(5')	C(1'), C(2')/C(4'), C(3')/C(5'), C(4')	H-C(3')/H-C(5')	C(1'), C(3')/ C(5'), C(4')	H-C(3')/ H-C(5')	C(1'), C(3')/ C(5'), C(4')
H-C(1")	COO	H-C(1") C(3)	H-C(1") C(3)	. ,	

signals due to  $\beta$ -D-glucose, a 4-(hydroxyphenyl)ethyl moiety, a tetrasubstituted benzene ring and a downfield-shifted Me group. The  $^{13}$ C-NMR ( $Table\ 2$ ), HSCQ, and HMBC experiments ( $Table\ 3$ ) allowed us to establish the structure of **4** as 1-{3-( $\beta$ -D-glucopyranosyloxy)-6-hydroxy-2-[2-(4-hydroxyphenyl)ethyl]phenyl}ethanone or tyrolobibenzyl C. Compound **4** represents a new natural compound and seems, in spite of the lack of a benzofuran ring system, to be biogenetically closely related to compounds **2** and **3**.

The  $^{13}$ C-NMR and DEPT spectrum (*Table 2*) of **4** showed twenty signals, two of them, at  $\delta$  129.7 (C(2'), C(6')) and 116.0 (C(3'), C(5')) with double intensity. Twelve of these signals, including the ones with double intensities, were the same as the corresponding signals of **2** and **3** and assignable to the (4-hydroxyphenyl)ethyl and the  $\beta$ -D-glucose moieties. Two of the remaining eight signals were attributable to an ethanone, *i.e.*  $\delta$  204.7 (CO) and 32.1 (Me), and the other six signals to the tetrasubstituted benzene ring.

The partial structures established by  ${}^{1}$ H- and  ${}^{13}$ C-NMR could be connected by means of the long-range correlations observed in the HMBC spectrum.  ${}^{1}$ H-NMR Multiplicities and coupling constants  $(J(4,5)=J(5,4)=9.0~{\rm Hz})$  of the aromatic H-C(4) and H-C(5) of the acetophenone moiety revealed that these protons had to be located vicinal to each other. The connection of  $\beta$ -D-glucose via an ether bond to C(3) of the acetophenone moiety was evidenced by the downfield shift of the anomeric-proton signal  $(\delta~4.90)$  and HMBC cross-peak with the C(3) signal. Two- and three-bond couplings between the signal of 1~H-C( $\alpha$ ) of the ethanediyl moiety and the signals of C(1), C(2), and C(3) localized the (4-hydroxyphenyl)ethyl moiety at C(2). The position of the ethanone group at C(1) was established by the HMBC cross-peak between the Me signal at  $\delta$  2.33 and that of C(1) at  $\delta$  131.0. Finally, three-bond couplings of H-C(4) with C(2) and C(6), as well as of H-C(5) with C(1) and C(3) revealed that the OH substituent was attached at C(6).

To verify the structure of compound **4**, it was acetylated to **4a**. The molecular mass of **4a** determined by ESI-MS as  $686 \ (m/z \ 686)$  showed the expected increase of 252 mass units in comparison with **4**. Additionally, <sup>1</sup>H-NMR (*Table 1*), <sup>13</sup>C-NMR (*Table 2*), HSQC, and HMBC (*Table 3*) data were in agreement with the proposed structure **4a**.

In a first attempt to evaluate the new natural products **2–4** for a potential pharmacological activity, we tested these substances, as well as the corresponding acetyl derivatives **3a** and **4a** and the known compound **1**, in the (<sup>3</sup>H)thymidine cytotoxicity assay against leukaemic cell lines GTB and HL60 [6]. None of the tested substances showed any cytotoxic effects in the concentration range between 0.25 and 4.00 μmol/l.

Pharmacological evaluations in other areas, like *e.g.* inflammation and antibiosis, are in progress. Furthermore, we are currently investigating the infrageneric distribution of this unique class of substances within the genus *Scorzonera* to obtain clues about their potential as chemotaxonomic markers.

The authors wish to thank Ms. M. Stütz for valuable technical assistance, Mag. pharm. D. Weigand for IR measurements, Prof. Dr. K.-H. Ongania and Dr. S. Sturm for MS measurements, as well as Dr. R. Hatschenberger and Prof. Dr. G. Konwalinka for cytotoxicity assays.

## **Experimental Part**

General. Column chromatography (CC): Merck silica gel 60 (40–63 μm). Semi-prep. HPLC: Merck 250 × 10 mm LiChrospher® RP-18 (10 μm material) column; isocratic MeCN/H<sub>2</sub>O 1:1, flow rate 3.5 ml/min; detection at 205 nm; for each run, injection of a 100-μl aliquot of a soln. containing 10 mg/ml of substance. FT-IR:  $\vec{v}_{\rm max}^{\rm ZnSe}$  in cm<sup>-1</sup>; microspectrometry. NMR Spectra: Varian-Unityplus-500 spectrometer at 500 (<sup>1</sup>H) and 125 MHz (<sup>13</sup>C) in CD<sub>3</sub>OD (1–4) or CDCl<sub>3</sub> (3a and 4a). MS: ESI, neg. mode; FAB, pos. mode; m/z (rel. %).

Plant Material. Scorzonera humilis L. was collected in the Leutasch valley near Unterkirchen, Tyrol, Austria, at an altitude of 1020 m. A voucher specimen is deposited in the herbarium of the Institute of Pharmacy. Extraction and Isolation. Partitioning of the MeOH extract (60.0 g) obtained from freeze-dried subaerial parts (190.8 g) of S. humilis between H<sub>2</sub>O/MeOH 2:1 and AcOEt yielded 18.4 g from the AcOEt phase.

Subsequent repeated CC (silica gel, gradients of petroleum ether/CH<sub>2</sub>Cl<sub>2</sub>/Me<sub>2</sub>CO, AcOEt/Me<sub>2</sub>CO, AcOEt/MeOH, and CH<sub>2</sub>Cl<sub>2</sub>/MeOH), CC (*Sephadex LH-20*, Me<sub>3</sub>OH/AcOEt/H<sub>2</sub>O (3:1:1), and consecutive CC (*Lobar RP18*, H<sub>2</sub>O/MeCN) yielded pure compounds **1** (17.2 mg), **2** (133.8 mg), **3** (158.2 mg), and **4** (93.4 mg).

Tumor Cell-Growth-Inhibition Assays. [ $^3$ H]Thymidine incorporation assays were performed as described in [6]. Briefly, 100- $\mu$ l aliquots of the cell suspensions (1 · 10 $^6$  cells/ml) were placed into wells of a microtiter plate and layered with 100  $\mu$ l of medium containing various concentrations of 1, 2, 3, 3a, 4, or 4a in a concentration range between 0.25 and 4.00  $\mu$ mol/l. Controls contained 100  $\mu$ l of pure medium without the test substances. Plates were then incubated for 48 h at 37 $^\circ$  under 5% CO<sub>2</sub> atmosphere and high humidity. All cultures were then pulsed for 24 h with 0.5 mCi of [ $^3$ H]thymidine per well. The samples were collected on glass-filter paper with a multiple automated harvester. The filters were dried at 55 $^\circ$  and then transferred to scintillation vials containing 4 ml of scintillation fluid. Radioactivity was measured in a liquid scintillation counter. Three replicate wells were used at each point, and experiments were performed in duplicate on different days (n = 6).

*Tyrolobibenzyl B* (3). White amorphous solid. M.p.  $136-146^{\circ}$  (dec.). FT-IR: 3400 (br.), 2927, 1670, 1612, 1568, 1514, 1458, 1382, 1249, 1176, 1075, 1029, 991. ESI-MS: 919 (100,  $[2M-H]^-$ ), 459 (12,  $[M-H]^-$ ), 297 (6,  $[M-glucose-H]^-$ ).

*Tyrolobibenzyl C* (4). White amorphous solid. M.p.  $118-122^{\circ}$  (dec.). FT-IR: 3400 (br.), 2935, 1678, 1611, 1514, 1485, 1453, 1357, 1258, 1075, 1040, 1019. ESI-MS: 433 (100,  $[M-H]^-$ ), 271 (90,  $M-glucose-H]^-$ ).

Acetylation of 3 and 4. To 3 (20 mg), aliquots of  $Ac_2O$  and pyridine were added. This soln, was mixed with cat, amounts of  $N_iN$ -dimethylpyridin-4-amine and kept at r.t. for 24 h. Subsequently, the mixture was separated by CC (silica gel, gradient  $CH_2Cl_2 \rightarrow MeOH$ ): 23.0 mg of an acetate mixture. The peracetate 3a was finally purified by isocratic semi-prep. HPLC ( $H_2O/MeCN$  1:1): 11.2 mg of pure peracetate 3a.

Likewise 4 was acetylated and the product purified: 38.0 mg of peracetate 4a.

2,3,4,6-Tetra-O-acetyl-β-D-glucopyranosyl 5-Acetoxy-4-[2-(4-acetoxyphenyl)ethyl]benzofuran-2-carboxyl-ate (3a). Colorless crystals. M.p. 231 – 245 $^{\circ}$  (dec.). FT-IR: 2924, 2854, 1733, 1644, 1465, 1377, 1368, 1250, 1170, 1148, 1097, 1028, 1013, 970. FAB-MS: 735 ( $[M+Na]^+$ , 100).

1-[6-Acetoxy-2-[2-(4-acetoxyphenyl)ethyl]-3-(2,3,4,6-tetra-O-acetyl- $\beta$ -D-glucopyranosyloxy)phenyl]ethanone (4a). Colorless crystals. M.p. 279 – 287° (dec.). FT-IR: 2949, 2880, 1763, 1749, 1704, 1598, 1508, 1465, 1430, 1372, 1242, 1194, 1068, 1045, 909. ESI-MS: 709 (100, [M + Na] $^+$ ), 379 (95, [M – (Ac) $_4$ glucose + Na] $^+$ ).

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